



MeOH much more readily than the iodides  $(\text{Me}_2\text{Si})_2\text{CSiR}_2\text{I}$  ( $\text{R} = \text{Ph}$  or  $\text{Me}$ ). This may be because the presence of a hydrogen atom permits more effective solvation of the forming cation.

We thank the S.R.C. for support, and Dow Corning Ltd., for gifts of chemicals.

(Received, 29th October 1980; Com. 1168.)

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